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Naming Covalent (Molecular) Compounds Quiz -Ouiziz Answer The world around you is made up of thousands and thousands of different compounds formed from chemical bonds There are three types of chemical bonds: Page 18/36

ionic bonding, covalent bonding and metallic bonding. This quiz will focus on covalent bonding. Use your knowledge of atoms and covalent bonding to answer the following questions. Group:

Chemical Bonding Page 19/36

II: Covalent Bonding Quiz Nice work! You're comfortable er naming covalent or molecular compounds and writing their formulas. If you're unsure of yourself, you can review the nomenclature rules and prefixes for covalent Page 20/36

compounds. From here, it's a good idea to know the properties of covalent compounds. How about another quiz?

Covalent Compound Names Quiz - ThoughtCo Compounds with Additional Page 21/36

Elements Quiz: Compounds with Additional FlementsSWer Introduction to Organic Compounds Quiz: Introduction to Organic Compounds Structural Formulas Quiz: Solids, Liquids, and Gases

Ouiz: Covalent Bonds Chapter 6 Ionic and Covalentswer Compound Naming (Practice Quiz) (with oxidation numbers and correct subscript latex codes) Take and pass with 70% for 5 point bonus on your test.

Chapter 6 Ionic and Covalentunds Compound Naming (Practice Quiz) Section Quiz: Covalent Bonding and Molecular Compounds In the space provided, write the letter of the term or phrase that best completes each statement or best Page 24/36

answers each question. 1 If two covalently bonded atoms move closer than a distance of the bond length, the potential energy of the atoms a becomes negative. b. decreases. increases

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Brearley High Schoolounds Chemical Compounds Ver Practice Quiz This online quiz is intended to give you extra practice in naming compounds, writing formulas and calculating molar masses (formula weights). Select Page 26/36

your preferences below and click 'Start' to give it a try! Z Answer

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Classify the Ca — Cl bond in CaCl 2 as ionic, polar covalent, orver nonpolar covalent. A ionic 5 The number of lone electron pairs in the NO 2 – ion is . A: 6 6. Which one of the following is most likely to be a covalent compound? A: SeO Page 35/36

27. Classify the O
— H bond in CH 3
OH as ionic, polar covalent, or nonpolar covalent.
A: polar ...

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