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Chem Worksheet 5-1 The behavior of
light indicates that it is comprised of waves

The distance between successive

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Chem Worksheet 5-1 The...

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Chem Worksheet 5-1 ...

In what part of the electromagnetic
spectrum are these waves? Light Waves
Name _____ Chem Worksheet 5-1
 $1 \text{ m} = 1 \times 10^3 \text{ mm}$
 $1 \text{ m} = 1 \times 10^6 \mu\text{m}$
useful equations
 $c = 3.00 \times 10^8 \text{ m/s}$
 $1 \text{ m} = 1 \times 10^9 \text{ nm}$
 $1 \text{ m} = 1 \times 10^{10} \text{ \AA}$
 $c =$

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3.00 $\times 10^8$ m/s Violet Light has shorter wavelength and higher frequency Red Light
Longer wavelength and lower frequency V
I B G Y O R 400 500 600 700 750 nm
wavelength (m) 10^{-12} 10^{-11} 10^{-10} 10^{-9} 10^{-8} 10^{-7} 10^{-6} 10^{-5} 10^{-4}
 10^{-3} 10^{-2} 10^{-1} 101 103 1018 1016 1014 1012 1010
108 106 104 10^{-11} am rays x-rays ultr ...

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Light Waves Name _____

Chem Worksheet 5-1. $1 \text{ m} = 1 \times 10^3 \text{ mm}$

$1 \text{ m} = 1 \times 10^6 \mu \text{ m}$. useful equations. $c =$

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! \times " $c = 3.00 \times 10^8 \text{ m/s}$ Shorter
wavelength (. $1 \text{ m} = 1 \times 10^9 \text{ nm}$ $1 \text{ m} = 1$
 $\times 10^{10} \text{ \AA}$! $c = 3.00 \times 10^8 \text{ m/s}$. Violet
Light. !) and higher frequency ("

1. Draw a wave below and label the
following parts: peak ...

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speed of light equation. $c=3.0 \times 10^8$ and
the equation with it is $c= \lambda \times \nu$ and can
be manipulated and isolate different parts
to get necessary answer. $\lambda =$ wave
length. energy equation. $e= h \times \nu$ where h

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is planks constant in JS of (6.626×10^{-34})
and V is frequency in -s and e is in J.

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5. What is the frequency of a 7.43×10^{-5} m wave?
6. What is the frequency of a

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2,600 cm wave? 7. What is the wavelength of a 4.34×10^{15} /s wave? 8. What is the frequency of a 2, 110 wave? 9. What is the frequency of a 5.6×10^{10} μ m wave? 10. What is the wavelength of 109.6 MHz wave? 11. What is the energy of a 7.66×10^{14} Hz wave? 12.

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Chemistry Worksheet – Wavelength,
frequency, & energy of ...

The diagram below shows white light dispersed into its different colours by a prism: Light travels at 300 000 km/s. It takes light 8.5 minutes to reach the Earth from the Sun. Apart from the electromagnetic waves, there are also

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matter waves, like water waves and sound waves. These travel at a much lower speed.

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Planck ' s Equation Name Chem

Worksheet 5-2. © John Erickson, 2005

WS5-2PlancksEq. $E = h \times \nu$ $h = 6.63 \times 10^{-34} \text{J} \cdot \text{s}$ useful equations $c = \lambda \times \nu$ $c = 3.00 \times 10^8 \text{m/s}$ $1 \text{ m} = 1 \times 10^9 \text{nm}$ $1 \text{ kJ} = 1000 \text{ J}$. example. Light with a wavelength of 525 nm is green. Calculate the energy in joules

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for a green light photon. - find the
frequency: $c = \lambda \cdot \nu$

Planck ' s Equation Name Chem
Worksheet 5-2

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Key Wwwwlightneasy within Wave Review
Worksheet Answers Pleasing Physical

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Answers Nature Of Worksheets Also Light
Waves throughout Wave Review
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