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CHAPTER 13 REVIEW Ions in Aqueous Solutions and Colligative
Properties MIXED REVIEW SHORT ANSWER Answer the
following questions in the space provided. 1. Match the four
compounds on the right to their descriptions on the left. b an ionic
compound that is quite soluble in water (a) HCl c an ionic
compound that is not very soluble in water (b) NaNO₃

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theoretically occupies zero volume. 0.08206 L atm/mol K. at
constant temperature, the volume of a given amount of gas i.... at
constant pressure, the volume of a given amount of gas is d....

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CHAPTER 6 REVIEW Chemical Bonding SECTION 3 SHORT
ANSWER Answer the following questions in the space provided. 1.
a The notation for sodium chloride, NaCl, stands for one (a)
formula unit. (c) crystal. (b) molecule. (d) atom. 2. d In a crystal of
an ionic compound, each cation is surrounded by a number of (a)
molecules. (c) dipoles. (b) ...

6 Chemical Bonding

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Test A, continued Use this figure to answer questions 7 and 8.

_____ 7. A solution containing 35 g of Li_2SO_4 dissolved in 100 g of water is heated from 10°C to 90°C . According to information in the figure, this temperature change would result in a. an additional 5 g of Li_2SO_4 in ...

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